

# **SAFETY DATA SHEET**

# SDS02754 HYDROGEN PEROXIDE 35%

Preparation Date: 17/Apr/2018

Version: 1

## 1. IDENTIFICATION

Product identifier

**Product Name** 

**HYDROGEN PEROXIDE 35%** 

Other means of identification

**SDS Number** 

SDS02754

**Synonyms** 

none

Recommended use of the chemical and restrictions on use

Recommended Use

Oxidizing agent Bleach & water chemicals.

Restricted Uses

No information available

Initial Supplier Identifier

Univar Canada Ltd. 9800 Van Horne Way Richmond, BC V6X 1W5 Telephone: 1-866-686-4827

Emergency telephone number

24 Hour Emergency Phone Number (CANUTEC): 1-888-226-8832 (1-888-CAN-UTEC)

## 2. HAZARD IDENTIFICATION

## Hazardous Classification of the substance or mixture

Oxidizing liquids	Category 2
Acute toxicity - Oral	Category 4
Acute toxicity - Inhalation (Dusts/Mists)	Category 4
Skin corrosion/irritation	Category 1
Sub-category A	
Serious eye damage/eye irritation	Category 1
Specific target organ toxicity (single exposure)	Category 3

## Label elements

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## **Hazard pictograms**



#### Signal Word: Danger

#### **Hazard statements**

May intensify fire; oxidizer
Harmful if swallowed
Harmful if inhaled
Causes severe skin burns and eye damage
May cause respiratory irritation

#### **Precautionary Statements**

#### Prevention

Wash face, hands and any exposed skin thoroughly after handling
Do not eat, drink or smoke when using this product
Use only outdoors or in a well-ventilated area
Do not breathe dust/fume/gas/mist/vapors/spray
Wear protective gloves/protective clothing/eye protection/face protection
Keep away from heat
Keep/Store away from clothing/ combustible materials
Take any precaution to avoid mixing with combustibles

#### Response

IF exposed or if you feel unwell: Call a POISON CENTER or doctor/physician

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

Continue rinsing

IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water or shower

Wash contaminated clothing before reuse

IF INHALED: Remove person to fresh air and keep comfortable for breathing

IF SWALLOWED: Rinse mouth. DO NOT induce vomiting

In case of fire: Use dry sand, dry chemical or alcohol-resistant foam to extinguish

#### **Storage**

Store locked up

Store in a well-ventilated place. Keep container tightly closed

#### **Disposal**

Dispose of contents/container to an approved waste disposal plant

Harmful to aquatic life with long lasting effects

Unknown acute toxicity

No information available

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## 3. COMPOSITION/INFORMATION ON INGREDIENTS

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#### Substance

Chemical Name	CAS No	Weight-%	Synonyms
Hydrogen Peroxide	7722-84-1	30 - 40%	Hydrogen Peroxide

## 4. FIRST AID

#### **Description of first aid measures**

#### General advice

Show this safety data sheet to the doctor in attendance. Immediate medical attention is required.

#### Inhalation

Remove to fresh air. If breathing has stopped, give artificial respiration. Get medical attention immediately. Do not use mouth-to-mouth method if victim ingested or inhaled the substance; give artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. If breathing is difficult, (trained personnel should) give oxygen. Delayed pulmonary edema may occur.

#### **Eve contact**

Rinse immediately with plenty of water, also under the eyelids, for at least 15 minutes. Keep eye wide open while rinsing. Do not rub affected area. Remove contact lenses, if present and easy to do. Continue rinsing. Get immediate medical advice/attention.

#### Skin contact

Wash off immediately with soap and plenty of water while removing all contaminated clothes and shoes. Get immediate medical advice/attention.

#### Ingestion

Do NOT induce vomiting. Clean mouth with water and drink afterwards plenty of water. Never give anything by mouth to an unconscious person. Get immediate medical advice/attention.

#### Self-protection of the first aider

Ensure that medical personnel are aware of the material(s) involved, take precautions to protect themselves and prevent spread of contamination. Avoid contact with skin, eyes or clothing. Avoid direct contact with skin. Use barrier to give mouth-to-mouth resuscitation. Wear personal protective clothing (see section 8).

## Most important symptoms and effects, both acute and delayed:

Corrosive Causes severe respiratory irritation. May cause conjunctivitis, corneal burns and permanent damage. Symptoms may occur with delay. Ingestion of high concentrations causes rapid release of oxygen which may expand the esophagus or stomach resulting in severe damage (bleeding, ulceration or perforation). Expected to cause burns to the gastrointestinal tract. Vapors may cause pulmonary edema. Toxic effects may be delayed. May cause burns resulting in permanent damage. Prolonged exposure may cause severe irritation and white discoloration. Burning may result in localized erythema (redness) or even blistering of the skin.

## Indication of any immediate medical attention and special treatment needed:

#### Note to physicians

Hydrogen peroxide at this concentration is a strong oxidant. Direct contact with the eye is likely to cause corneal damage especially if not washed immediately. Careful ophthalmologic evaluation is recommended and the possibility of local corticosteroid therapy should be considered. Because of the likelihood of corrosive effects on the gastrointestinal tract after ingestion, and the unlikelihood of systemic effects, attempts at evacuating the stomach via emesis induction or gastric lavage should be avoided. There is a remote possibility, however, that a nasogastric or orogastric tube may be required for the reduction of severe distension due to gas formation.

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#### 5. FIRE-FIGHTING MEASURES

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#### Suitable Extinguishing Media

Do not use CO2 extinguisher on this material; use only water spray or appropriate foam. Do not use organic compounds on this material.

#### Specific hazards arising from the substance or mixture

Stay upwind. Isolate and restrict area access. Stop leak only if safe to do so. Fight fire from a safe distance and from a protected location. Containers exposed to intense heat from fires should be cooled with water to prevent vapor pressure build-up which could result in container rupture. Strong oxidizer. Contact with combustible materials may cause a fire. Release of oxygen may support combustion. Contact with incompatible materials (e.g. metals, alkalis and reducing agents) will cause hazardous decomposition resulting in the release of large quantities of heat, steam and oxygen gas. Exposure to heat may cause hazardous decomposition. A severe detonation hazard may exist when mixed with organic liquids, e.g. kerosene or gasoline.

#### Hazardous combustion products

Oxygen. Steam.

#### Special protective equipment for fire-fighters

Firefighters should wear self-contained breathing apparatus and full firefighting turnout gear. Use personal protection equipment.

### 6. ACCIDENTAL RELEASE MEASURES

#### Personal precautions, protective equipment and emergency procedures

Attention! Corrosive material. Avoid contact with skin, eyes or clothing. Ensure adequate ventilation. Use personal protective equipment as required. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

#### Environmental precautions

Prevent further leakage or spillage if safe to do so. Should not be released into the environment. Do not allow to enter into soil/subsoil. Prevent product from entering drains.

#### Methods and materials for containment and cleaning up

Prevent further leakage or spillage if safe to do so.

### 7. HANDLING AND STORAGE

## Precautions for safe handling

For food plant and other industrial use only. Handle and open containers with care. Never touch eyes of face with hands or gloves that may be contaminated with this product. Do not ingest. Avoid inhalation of chemical. Empty containers may contain hazardous product residues. Keep the containers closed when not in use. Protect against physical damage. Use appropriate personnel protective equipment.

#### Conditions for safe storage, including any incompatibilities

Keep containers tightly closed. Store in a cool, dry, well ventilated area. Do not store near combustible materials. Do not store this material in containers made of light metals. Recommended container materials: glass, polyvinyl chloride, polyethylene, ceramics, polypropylene. Use adequate venting devices on all packages, containers and tanks and check correct operation periodically. Do not confine product in unvented vessels or between closed valves. Risk of overpressure and bursting due to decomposition in confined spaces and pipes. Do not store on wooden floors or

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wooden pallets. Protect from direct sunlight and heat.

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

#### Control parameters

#### **Exposure Limits**

Chemical Name	Alberta OEL	British Columbia OEL	Ontario	Quebec OEL	Exposure Limit - ACGIH	Immediately Dangerous to Life or Health - IDLH
Hydrogen Peroxide 7722-84-1	TWA: 1 ppm TWA: 1.4 mg/m <sup>3</sup>	TWA: 1 ppm	TWA: 1 ppm	TWA: 1 ppm TWA: 1.4 mg/m <sup>3</sup>	1 ppm TLV-TWA	

Consult local authorities for recommended exposure limits

## Appropriate engineering controls

#### Engineering controls

Use process enclosure, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits.

## Individual protection measures, such as personal protective equipment

#### Eye/face protection

Face protection shield.

#### Hand protection

Butyl rubber gloves. Nitrile gloves. Natural rubber gloves.

#### Skin and body protection

Wear suitable protective clothing. Long sleeved clothing. Chemical resistant apron.

#### Respiratory protection

If exposure exceeds occupational exposure limits, use an appropriate NIOSH-approved respirator.

### General hygiene considerations

Avoid contact with skin, eyes or clothing. Wear suitable gloves and eye/face protection. Do not eat, drink or smoke when using this product. Remove and wash contaminated clothing and gloves, including the inside, before re-use. Contaminated work clothing should not be allowed out of the workplace. Regular cleaning of equipment, work area and clothing is recommended. Wash hands before breaks and immediately after handling the product.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

## Information on basic physical and chemical properties

**Appearance** 

**Physical state** 

Liquid

Color

Clear Colorless

Odor

Pungent

**Odor threshold** 

No information available

**PROPERTIES** 

pH

**Values** <2 (20°C) Remarks • Method

Melting point / freezing point -56 °C / -69 °F

Initial boiling point/boiling range 119 °C / 246 °F

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Flash point No data available None known
Evaporation rate No data available None known
Flammability (solid, gas) No data available None known

Flammability Limit in Air
Upper flammability limit: No data available

Lower flammability limit: > 40

Vapor pressure 48 Pa @ 30°C
Relative vapor density No data available

Specific Gravity 1.13

Water solubility
Solubility in other solvents
Partition coefficient
Autoignition temperature
Completely miscible
No data available
No data available
No data available

Autoignition temperature

Decomposition temperature

Kinematic viscosity

No data available
No data available
1.8 mPa.s @ 0°C

Kinematic viscosity

1.8 mPa.s @ 0°C

No data available

**Explosive properties**No information available. **Oxidizing properties**No information available.

Molecular weight

VOC Percentage Volatility Liquid Density Bulk density 34.02 g/Mol

No information available No information available No information available

## 10. STABILITY AND REACTIVITY

None known

None known

None known

None known

## Reactivity/Chemical Stability

Stable

## Possibility of hazardous reactions

No additional remark.

#### **Hazardous polymerization**

Will not occur.

#### Conditions to avoid

High temperatures. Spontaneous combustion hazard: - Combustible materials exposed to hydrogen peroxide should be immediately submerged in or rinsed with large amounts of water to ensure that all hydrogen peroxide is removed. Residual hydrogen peroxide that is allowed to dry (upon evaporation hydrogen peroxide can concentrate) on organic materials such as paper, fabrics, cotton, leather, wood, or other combustibles, can cause the material to ignite and result in a fire.

#### Incompatible materials

Organic materials. Reducing agents. Alkalis. Combustible material. Metals. Heavy metals and their salts.

#### Hazardous decomposition products

Oxygen. Steam.

## 11. TOXICOLOGICAL INFORMATION

#### Information on likely routes of exposure

#### **Inhalation**

Causes severe respiratory irritation. Vapors may cause pulmonary edema. Toxic effects may be delayed.

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## Eve contact

Corrosive. May cause conjunctivitis, corneal burns and permanent damage. Symptoms may occur with delay.

#### Skin contact

Corrosive. May cause burns resulting in permanent damage. Prolonged exposure may cause severe irritation and white discoloration. Burning may result in localized erythema (redness) or even blistering of the skin.

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#### Ingestion

Ingestion of high concentrations causes rapid release of oxygen which may expand the esophagus or stomach resulting in severe damage (bleeding, ulceration or perforation). Expected to cause burns to the gastrointestinal tract.

### Information on toxicological effects

#### **Symptoms**

Redness. Burning. May cause blindness. Coughing and/ or wheezing.

#### Numerical measures of toxicity

#### **Acute toxicity**

## The following values are calculated based on chapter 3.1 of the GHS document

ATEmix (oral)

1,518.00 mg/kg

**ATEmix** 

1.50 mg/l

(inhalation-dust/mist)

#### **Unknown acute toxicity**

No information available

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50
Hydrogen Peroxide 7722-84-1	= 1518 mg/kg ( Rat )	= 9200 mg/kg ( Rabbit )	= 2000 mg/m³ (Rat) 4 h

#### Delayed and immediate effects as well as chronic effects from short and long-term exposure

### Skin corrosion/irritation

Corrosive. May cause burns resulting in permanent damage. Prolonged exposure may cause severe irritation and white discoloration. Burning may result in localized erythema (redness) or even blistering of the skin.

## Serious eye damage/eye irritation

Corrosive. May cause conjunctivitis, corneal burns and permanent damage. Symptoms may occur with delay.

#### Respiratory or skin sensitization

No information available.

### Germ cell mutagenicity

No information available.

### Carcinogenicity

No information available.

The table below indicates whether each agency has listed any ingredient as a carcinogen.

Chemical Name	ACGIH	IARC	NTP	OSHA
Hydrogen Peroxide 7722-84-1	A3	Group 3	Not available	Not available

#### Legend

## ACGIH (American Conference of Governmental Industrial Hygienists)

A3 - Animal Carcinogen

#### IARC (International Agency for Research on Cancer)

Group 3 - Not Classifiable as to Carcinogenicity in Humans

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Reproductive toxicity

It is not possible to conclude that hydrogen peroxide is mutagenic. Positive results have been obtained in cultured humans cells. Negative results have been obtained in relevant studies using live animals. Positive results have been obtained in short-term mutagenicity tests.

Specific target organ systemic toxicity - single exposure

May cause respiratory irritation.

Specific target organ systemic toxicity - repeated exposure

No information available.

**Aspiration hazard** 

No information available.

## 12. ECOLOGICAL INFORMATION

## **Ecotoxicity**

Chemical Name	Ecotoxicity - Freshwater Algae Data	Ecotoxicity - Fish Species Data	Toxicity to microorganisms	Crustacea
Hydrogen Peroxide 7722-84-1	Not available	16.4 mg/L LC50 (Pimephales promelas) 96 h 18 - 56 mg/L LC50 (Lepomis macrochirus) 96 h static 10.0 - 32.0 mg/L LC50 (Oncorhynchus mykiss) 96 h static	Not available	EC50: 18 - 32mg/L (48h, Daphnia magna)

Persistence and degradability

No information available.

Bioaccumulation

No information available.

Chemical Name	Partition coefficient
Hydrogen Peroxide	Not available
7722-84-1	

Other adverse effects

No information available.

## 13. DISPOSAL CONSIDERATIONS

#### Waste treatment methods

Dispose of in accordance with local regulations. Dispose of waste in accordance with environmental legislation.

Do not reuse empty containers.

### 14. TRANSPORT INFORMATION

TDG (Canada):

UN Number

UN2014

Shipping name Class

HYDROGEN PEROXIDE, AQUEOUS SOLUTION

5.1 (8)

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**Packing Group** 

Ш

Marine pollutant

Not available.

DOT (U.S.)

**UN Number** 

UN2014

Shipping name

HYDROGEN PEROXIDE, AQUEOUS SOLUTION

Class

5.1 (8)

**Packing Group** 

Ш

Marine pollutant

Not available

## 15. REGULATORY INFORMATION

Safety, health and environmental regulations/legislation specific for the substance or mixture

**NSF** International



#### Additional information

Maximum use for potable water 3 mg/L. Only products bearing the NSF Mark on the product, product packaging, and/or documentation shipped with the product are Certified.

**U.S. Regulatory Rules** 

Chemical Name	CERCLA/SARA - Section 302:	SARA (311, 312) Hazard Class:	CERCLA/SARA - Section 313:
Hydrogen Peroxide - 7722-84-1	Listed	Not Listed	Not Listed

International Inventories

**TSCA DSL/NDSL**  Complies Complies

Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

## 16. OTHER INFORMATION, INCLUDING DATE OF PREPARATION OF THE LAST REVISION

NFPA:

Health hazards 3

Flammability 0

Instability 0

Physical and

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HMIS Health Rating: Health hazards 3

Flammability 0

Physical hazards 0

chemical properties -Personal protection

Legend Section 8: EXPOSURE CONTROLS/PERSONAL PROTECTION

TWA

TWA (time-weighted average)

STEL

STEL (Short Term Exposure Limit)

Ceiling

Maximum limit value

Skin designation

Prepared By:

The Environment, Health and Safety Department of Univar Canada Ltd.

**Preparation Date: Revision Date:** 

17/Apr/2018 17/Apr/2018

Disclaimer

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**End of Safety Data Sheet**